

GENERIC MARKING PRINCIPLE 5:

Marks should be awarded using the full range of marks defined in the mark scheme for the question (however; the use of the full mark range may be limited according to the quality of the candidate responses seen).

GENERIC MARKING PRINCIPLE 6:

Marks awarded are based solely on the requirements as defined in the mark scheme. Marks should not be awarded with grade thresholds or grade descriptors in mind.

Science-Specific Marking Principles

- 1 Examiners should consider the context and scientific use of any keywords when awarding marks. Although keywords may be present, marks should not be awarded if the keywords are used incorrectly.
- 2 The examiner should not choose between contradictory statements given in the same question part, and credit should not be awarded for any correct statement that is contradicted within the same question part. Wrong science that is irrelevant to the question should be ignored.
- 3 Although spellings do not have to be correct, spellings of syllabus terms must allow for clear and unambiguous separation from other syllabus terms with which they may be confused (e.g. ethane / ethene, glucagon / glycogen, refraction / reflection).
- 4 The error carried forward (ecf) principle should be applied, where appropriate. If an incorrect answer is subsequently used in a scientifically correct way, the candidate should be awarded these subsequent marking points. Further guidance will be included in the mark scheme where necessary and any exceptions to this general principle will be noted.
- 5 'List rule' guidance
For questions that require *n* responses (e.g. State **two** reasons ...):
 - The response should be read as continuous prose, even when numbered answer spaces are provided.
 - Any response marked *ignore* in the mark scheme should not count towards *n*.
 - Incorrect responses should not be awarded credit but will still count towards *n*.
 - Read the entire response to check for any responses that contradict those that would otherwise be credited. Credit should **not** be awarded for any responses that are contradicted within the rest of the response. Where two responses contradict one another, this should be treated as a single incorrect response.
 - Non-contradictory responses after the first *n* responses may be ignored even if they include incorrect science.

6 Calculation specific guidance

Correct answers to calculations should be given full credit even if there is no working or incorrect working, **unless** the question states 'show your working'.

For questions in which the number of significant figures required is not stated, credit should be awarded for correct answers when rounded by the examiner to the number of significant figures given in the mark scheme. This may not apply to measured values.

For answers given in standard form (e.g. $a \times 10^n$) in which the convention of restricting the value of the coefficient (a) to a value between 1 and 10 is not followed, credit may still be awarded if the answer can be converted to the answer given in the mark scheme.

Unless a separate mark is given for a unit, a missing or incorrect unit will normally mean that the final calculation mark is not awarded. Exceptions to this general principle will be noted in the mark scheme.

7 Guidance for chemical equations

Multiples / fractions of coefficients used in chemical equations are acceptable unless stated otherwise in the mark scheme.

State symbols given in an equation should be ignored unless asked for in the question or stated otherwise in the mark scheme.



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Question	Answer	Marks
1(a)(i)	Value of L in range 43.0 to 47.0 cm, with unit.	1
1(a)(ii)	Value of h less than H and θ to nearest degree and in range 5° to 40° .	1
1(b)	Six sets of readings of h and θ with correct trend and without help scores 4 marks, five sets scores 3 marks etc.	4
1(b)	Range: $\theta_{\min} \leq 10^\circ$ and $\theta_{\max} \geq 30^\circ$	1
1(b)	Column headings: Each column heading must contain a quantity and a unit where appropriate The presentation of quantity and unit must conform to accepted scientific convention e.g. $\theta/^\circ$ $\sin\theta$ must not have a unit.	1
1(b)	Consistency: All raw values of h must be given to the nearest mm	1
1(b)	Significant figures: Values of $\sin\theta$ given to 3 s.f.	1
1(b)	Calculation: Values of $\sin\theta$ calculated correctly	1
1(c)(i)	Axes: Sensible scales are used, no awkward scales (e.g. 3:10) Scales are chosen so that the plotted points occupy at least half the graph grid in both x and y directions Scales are labelled with the quantity which is being plotted. Scale markings are no more than 2 cm apart.	1
1(c)(i)	Plotting of points: All observations from the table are plotted on the grid. Diameters of plotted points are $<$ half a small square (no blobs). Plots are accurate to within half a small square in both x and y directions.	1



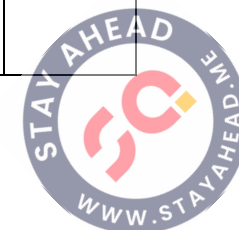
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Question	Answer	Marks
1(c)(i)	Quality: All points in the table must be plotted (at least 5) for this mark to be awarded. It must be possible to draw a straight line that is within ± 0.02 on the $\sin \theta$ axis of all plotted points. The trend must be negative.	1
1(c)(ii)	Line of best fit: Judged by balance of all points on the grid (at least 5) about the candidate's line. There must be an even distribution of points either side of the line along the full length If all points lie on trend candidates do not need to indicate an anomalous point. However, a single off-trend point is allowed as anomalous if it is circled or labelled and there are at least five other plotted points. Lines must not be kinked or thicker than half a square.	1
1(c)(iii)	Gradient: The hypotenuse of the triangle used must be greater than half the length of the drawn line. Method of calculation must be correct. Not $\Delta x / \Delta y$. Both read-offs must be accurate to half a small square in both the x and y directions.	1
1(c)(iii)	y -intercept: Correct read-off from a point on the line substituted into $y = mx + c$ or an equivalent expression, with read-off accurate to half a small square in both x and y directions. or Intercept read directly from the graph, with read-off at $h = \text{zero}$ accurate to half a small square in y direction.	1
1(d)	a equal to candidate's gradient and b equal to candidate's intercept. Values must not be written as fractions or to only 1 significant figure.	1
1(d)	Units for a and b correct and consistent with value (e.g. cm^{-1} for a and no unit for b)	1
1(e)	Correct calculation of M	1



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Question	Answer	Marks
2(a)	Raw values for v to nearest mm and final value in range 39.0 to 41.0 cm.	1
2(a)	Final value for u_0 in range 4.0 to 6.0 cm.	1
2(b)(i)	Raw values for u to nearest mm.	1
2(b)(ii)	Absolute uncertainty of 0.5 to 1.0 cm and correct method of calculation to obtain percentage uncertainty in u . If several readings have been taken, then the absolute uncertainty can be half the range, but not zero if values are equal.	1
2(b)(iii)	Correct calculation of f	1
2(b)(iii)	Correct calculation of n	1
2(b)(iv)	Justification correctly linked to the significant figures in v and u	1
2(c)	Second value of v	1
2(c)	Second value of u	1
2(c)	Quality: second u greater than first u	1
2(d)	Two values of k calculated correctly Final values of k are not written as fractions	1
2(e)	Calculation of percentage difference between candidate's two k values. Comparison of percentage difference with 20% leading to a consistent conclusion.	1
2(f)(i)	A Two k values are not enough to draw a valid conclusion B Difficult to judge when image is sharpest / difficult to produce a sharp image C Difficult to hold torch in line with lens (axis) D Difficult to measure u (or v) due to parallax error / deciding lens position E Second solution may be diluted by traces of first solution / some of first solution remains on the lens F LEDs not at front of torch <i>1 mark for each point up to a maximum of 4.</i>	4



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Question	Answer	Marks
2(f)(ii)	A Take more readings and plot a graph / calculate more k values and compare B Method to reduce unwanted light e.g. use screens around the area of the image / use darkroom C Mount torch on block (or stand) OR use an optical bench D Use set square next to rule (as pointer) / mark lens position on container E Clean the lens / clean the container / use a new container for second test F Remove glass from front of torch <i>1 mark for each point up to a maximum of 4.</i>	4

